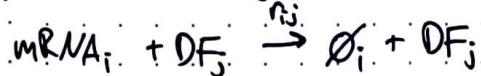


Transfer learning: Degradation Factor Concentration

Equation for degradation of mRNA_i with degradation factor DF_j:



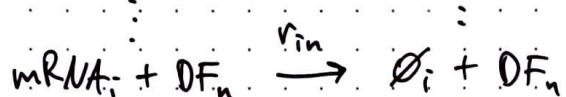
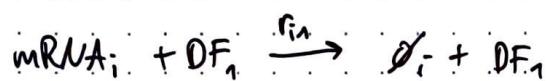
with reaction rate $r_{ij} = k_{ij} \cdot \text{DF}_j \cdot \text{mRNA}_i$, where

k_{ij} ~ binding probability of mRNA_i and DF_j

DF_j ~ concentration of DF_j

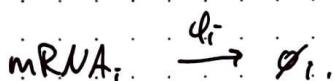
mRNA_i ~ concentration of mRNA_i

The half-life of an mRNA_i is determined by all the r_{ij} 's in combination:



Now, let $\varphi_{ij} = k_{ij} \cdot \text{DF}_j$ and consider $r_{ij} = \varphi_{ij} \cdot \text{mRNA}_i$.

Then $\varphi_i := \sum_j \varphi_{ij}$ leads to the simplification



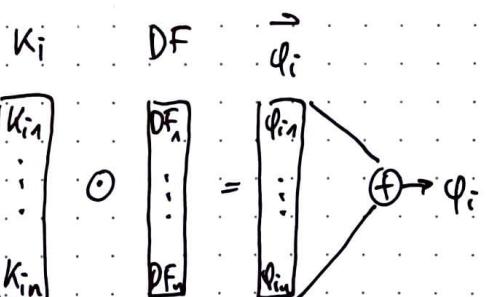
where $\frac{1}{\varphi_i} \sim \text{Half-life of mRNA}_i$.

We can try to model this simplification in order to learn a metric of concentration of the different DF's across different tissues:

$$\varphi_{ij} = k_{ij} \cdot \text{DF}_j$$

↑ ↑ ↑

proportional to obtained from learned
experimentally previous parameters
obtained half-life



for each tissue-type and each mRNA_i.

$$\text{HL}_i \sim \frac{1}{\varphi_i}$$

$$\text{with } \varphi_i = \sum_j \varphi_{ij}$$